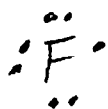
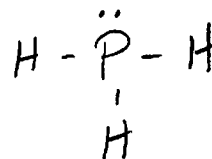


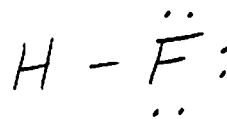
1. Fluorine (neutral atom)

5. Phosphorus trihydride (PH<sub>3</sub>)

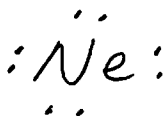
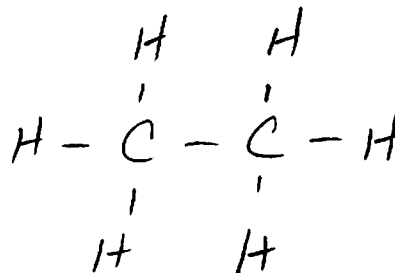
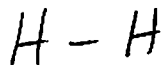
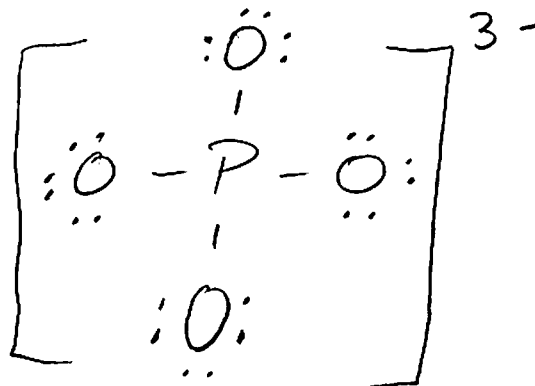
2. Helium



6. Hydrofluoric acid (HF)

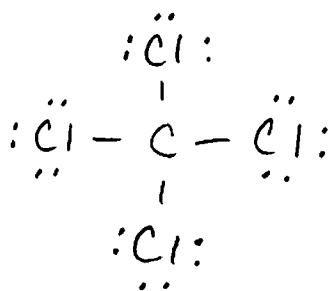


3. Neon

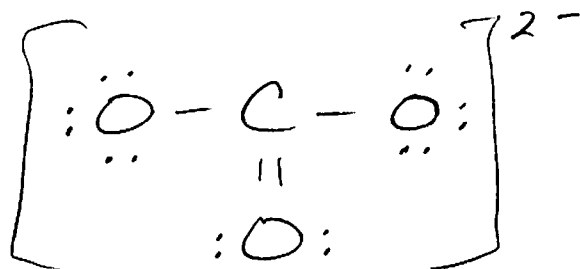
7. Ethane (C<sub>2</sub>H<sub>6</sub>)4. Diatomic hydrogen (H<sub>2</sub>)8. Phosphate ion (PO<sub>4</sub><sup>3-</sup>)

Practice Problems: Lewis Dot Structures (Page 2)

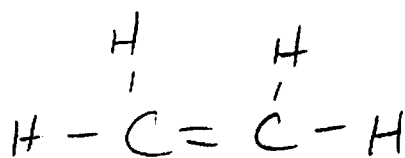
9. Carbon tetrachloride



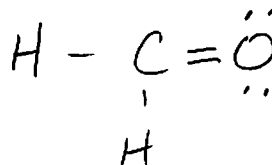
13. Carbonate ion



10. Ethene (C<sub>2</sub>H<sub>4</sub>)



14. Formaldehyde (CH<sub>2</sub>O)



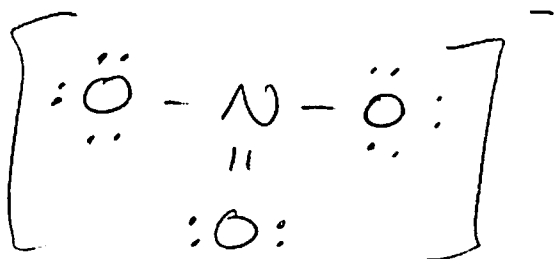
11. Carbon dioxide



15. Hydrogen cyanide (HCN)



12. Nitrate ion



16. Carbon monoxide

